

UN FUTURO SENZA IL PETROLIO:

IL RUOLO FONDAMENTALE DELLA CHIMICA

(ALCUN ESPERIENZE NEL CAMPO DEI NUOVI MATERIALI)

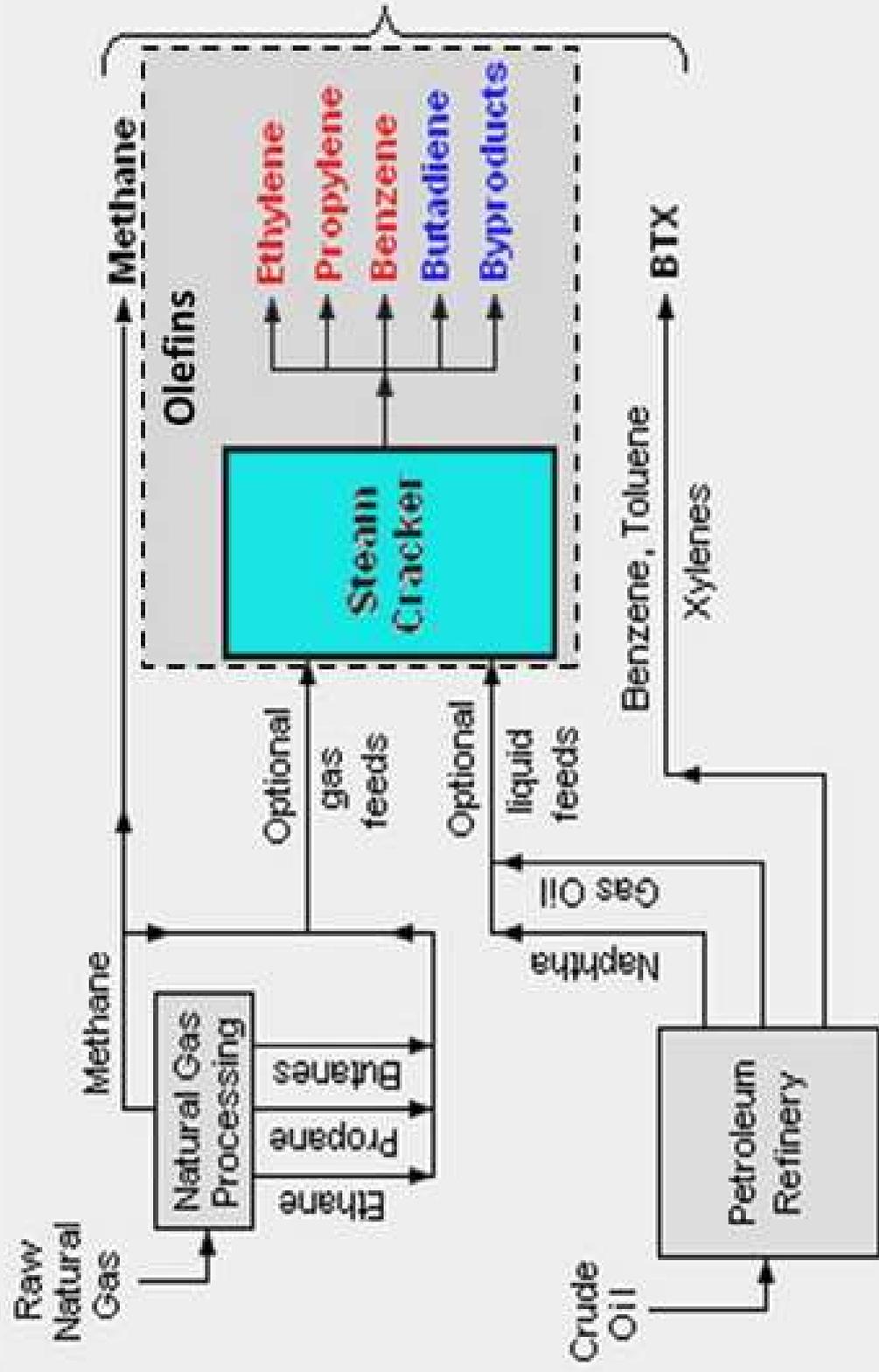
G. Chidichimo

World Petroleum Energy Crisis

By John Bachar (jmbachar@sbcglobal.net)

Professor Emeritus of Mathematics from California State University, member of SCFS (Southern California Federation of Scientists) Executive Board.

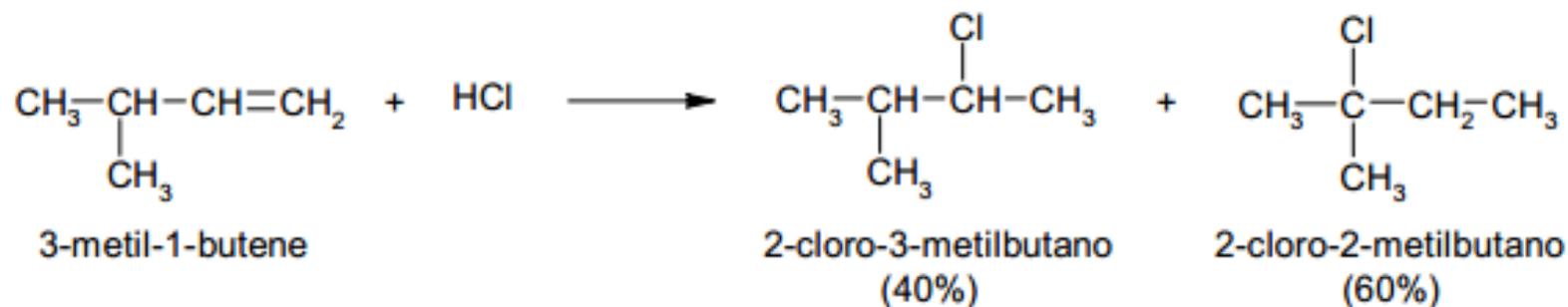
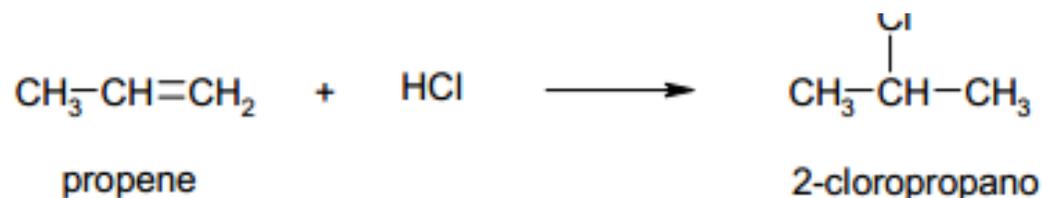
Only 1.08 trillion barrels of petroleum reserves are left on Earth (World Oil Journal 2005), and only one new barrel is found for every four used. With over 31.03 billion barrels consumed annually worldwide (2006) and rapidly increasing, **there is less than 25.4 years left.** Then it's all gone! Of the 48 petroleum-producing countries, only 15 can supply their own internal needs; the remaining 33 must supplement their internal needs with imports.

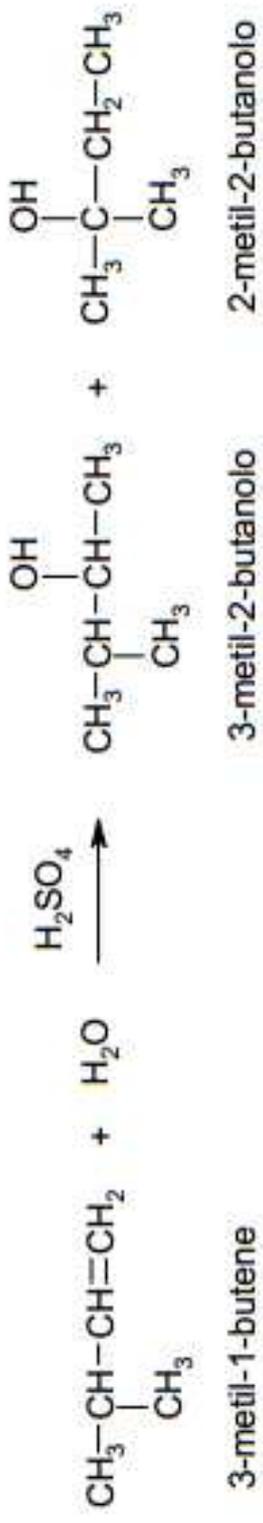
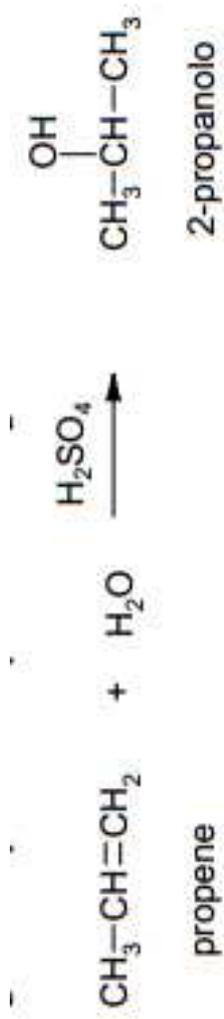


Petrochemical intermediate feedstocks

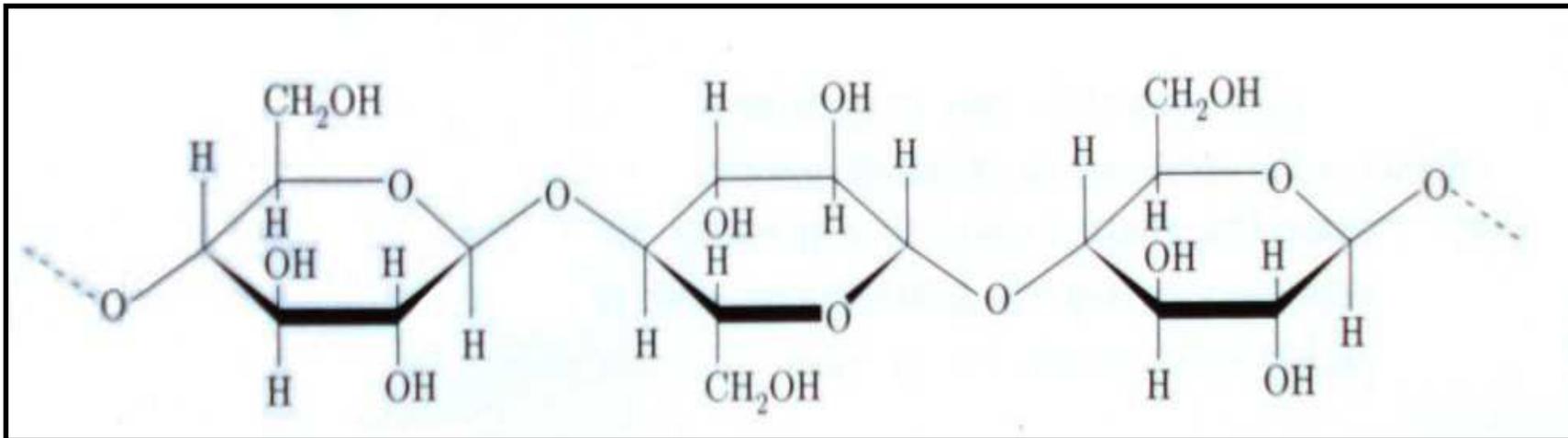
Produced by cracking any of the optional feeds
 Produced only by cracking any of the liquid feeds

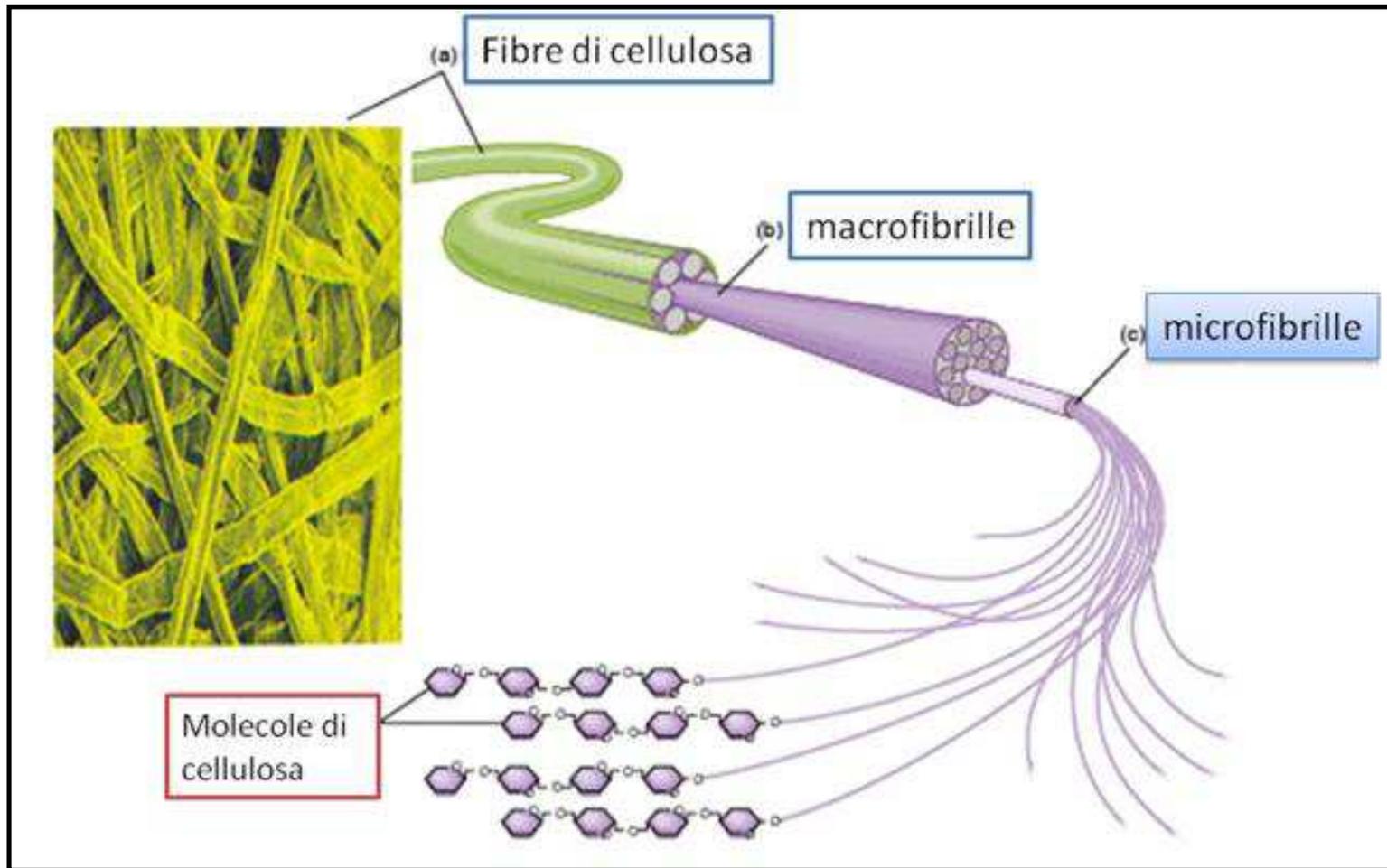
LA PRODUZIONE DI OLEFINE DAL PETROLIO E' L'INIZIO DELLA VARIEGATA CATENA DI PRODOTTI CHE DAL PETROLIO PRENDE ORIGINE.





La celulosa





Struttura delle fibre di cellulosa

La ginestra non è solo bella!



La canna è un'altra risorsa



FIBRA TESSILE MA E' CELLULOSA!



Our Patents

Chidichimo G. , Alampi C. , Cerchiara T. , Gabriele B. , Salerno G. , Vetere M. , *Physical chemical process for production of vegetable fibers*. Depositario: Università della Calabria. *vedi documento originale World Patent*, **WO2007102184A2 2007**.

G. Chidichimo, A. Aloise, M. De Benedittis, A. De Rango, G. Esposito, V. Gallo, S. Manfredi, G. Pingitore , *Materiali compositi ottenuti da fibre estratte da due varietà di ginestra*, **CS 2013°000027**

G. Chidichimo, A. Aloise, M. De Benedittis, A. De Rango, G. Esposito, S. Manfredi, G. Pingitore, *Materiali compositi ottenuti da matrici vegetali di ginestra e sostanze collanti di tipo organico ed inorganico; e processi per la loro realizzazione*, **In application**

G. Danieli, G. Chidichimo, P. F. Greco, P. Nudo, A. Aloise e A. De Rango, *Impianto automatizzato per l'estrazione di fibre vegetali da piante*, **In application**

Microfibres for new plastics



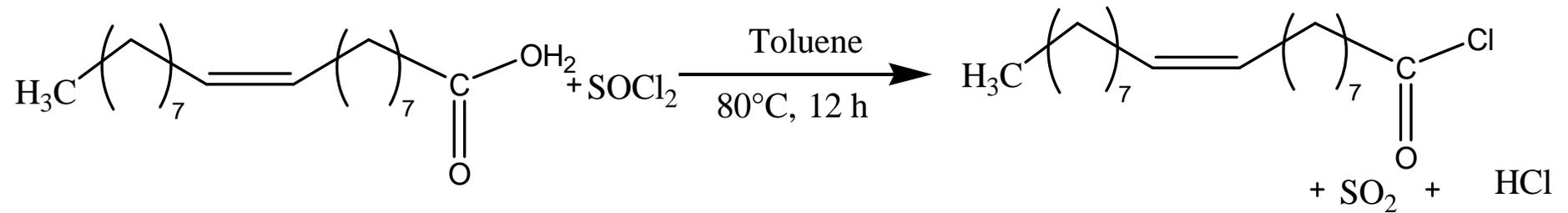
Arundo Donax



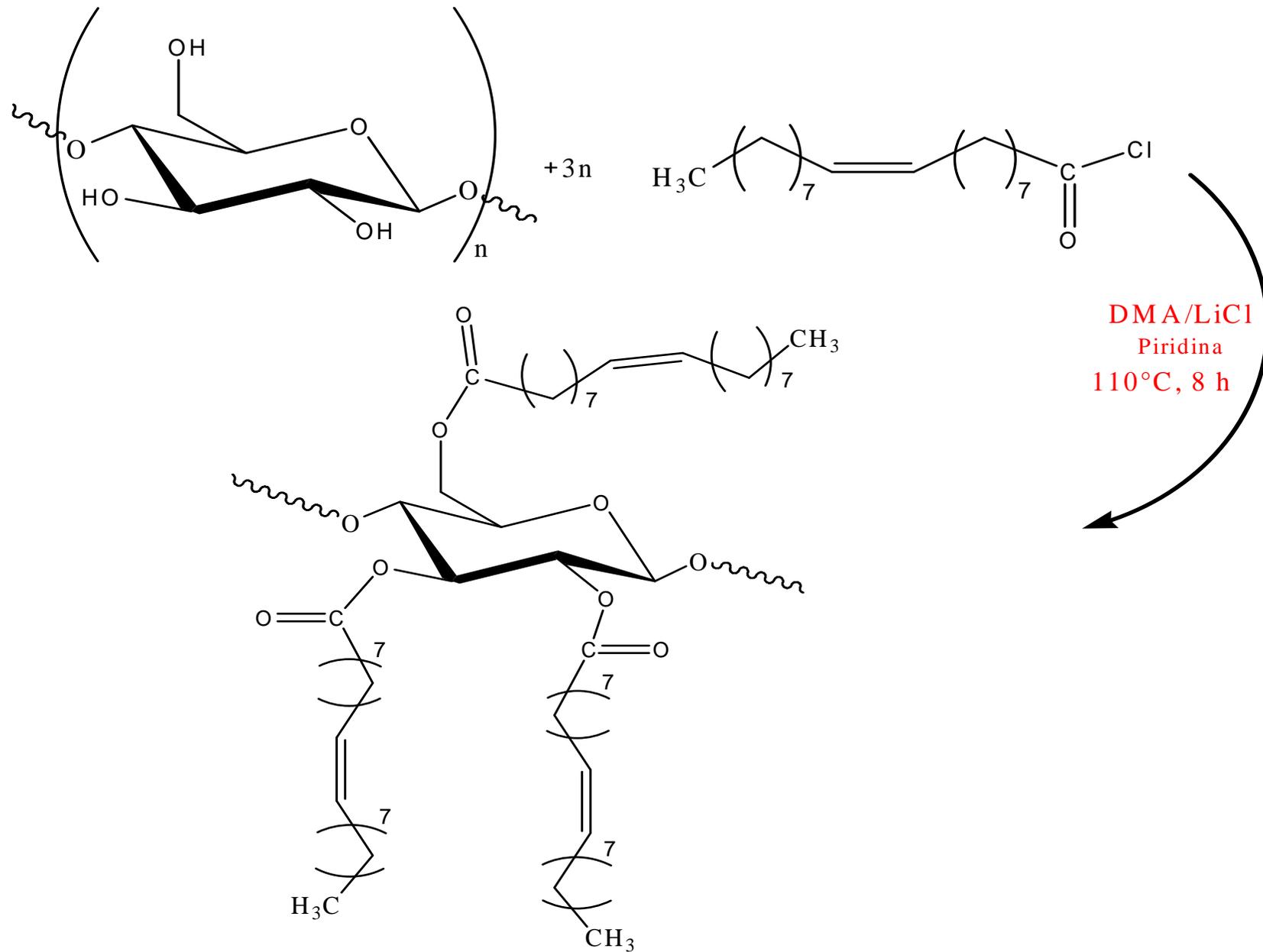
Spartium Juncem

Produzione di compositi di fibra cellulosica con PP e PE

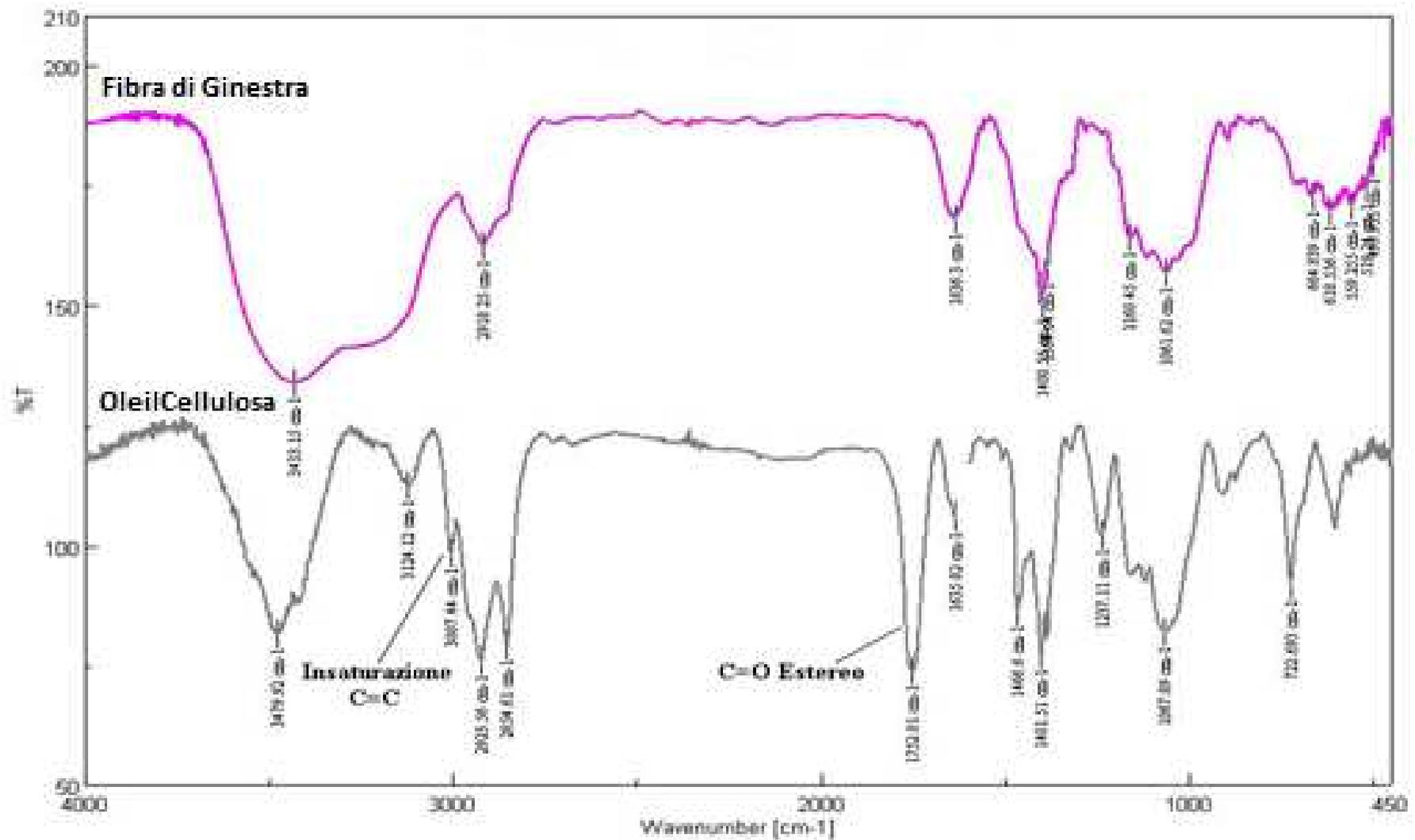
1) Sintesi dell'oleil-cloruro :



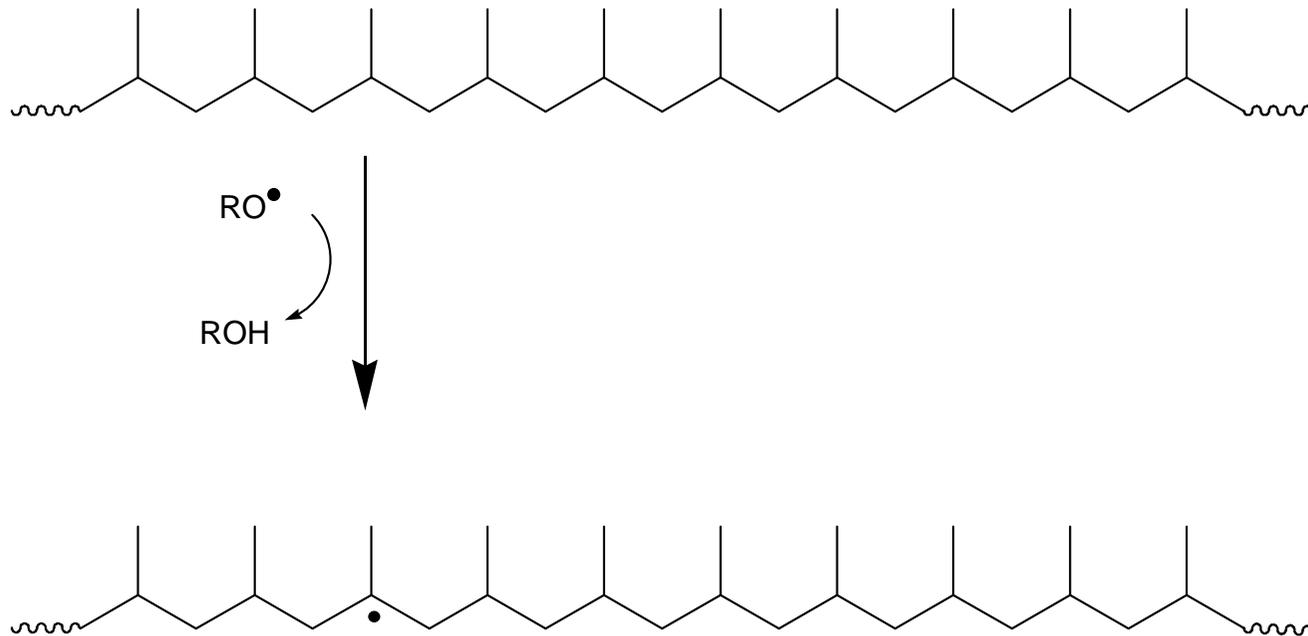
2) Esterificazione della fibra:

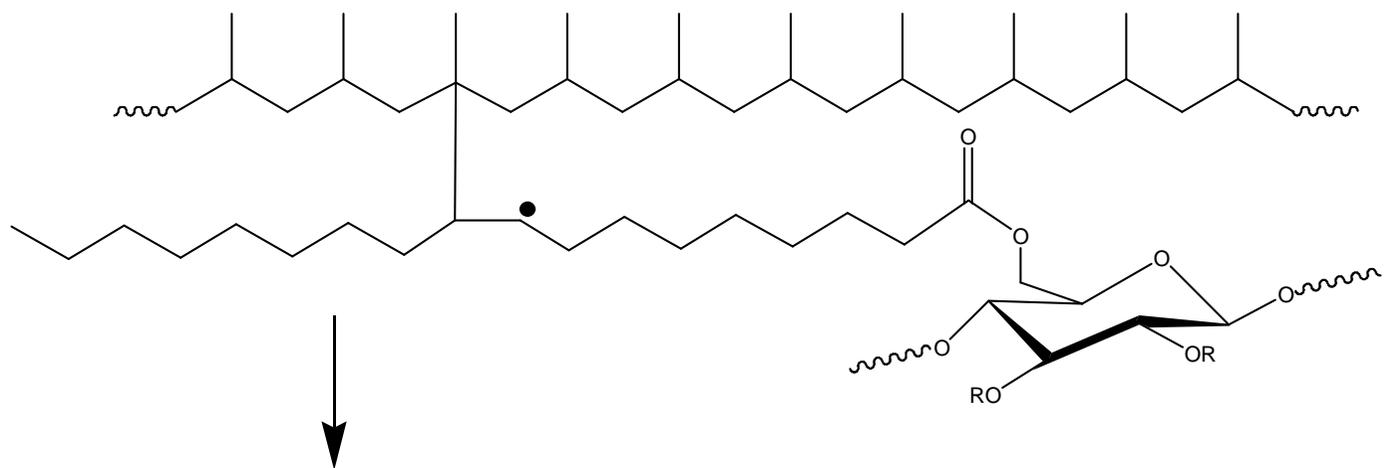
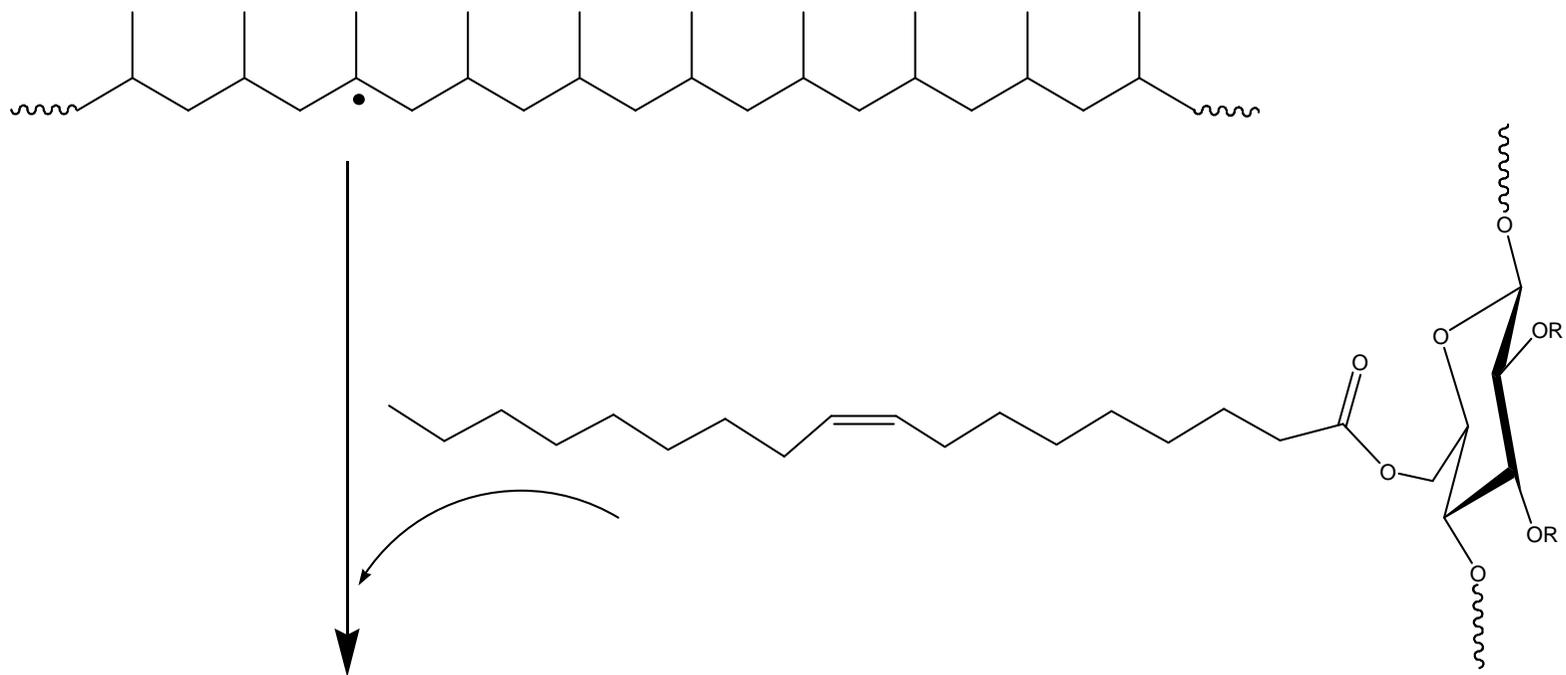


Confronto tra gli spettri IR della fibra pura e della fibra esterificata:

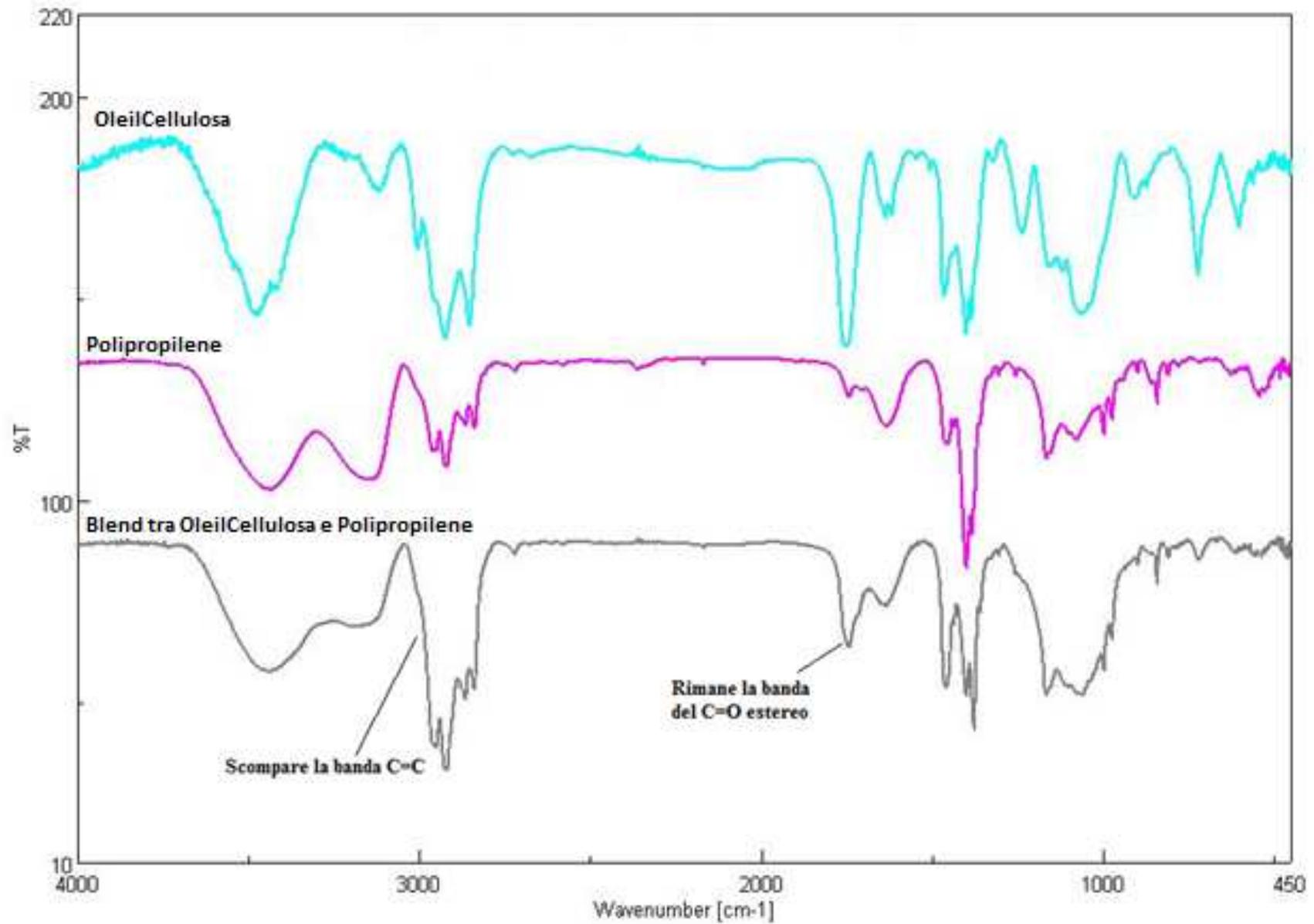


PREPARAZIONE DI UN BLEND POLIMERICO INNOVATIVO TRA OLEILCELLULOSA E POLIPROPILENE





Spettri IR dell' OleilCellulosa, del Polipropilene e del nuovo blend formato



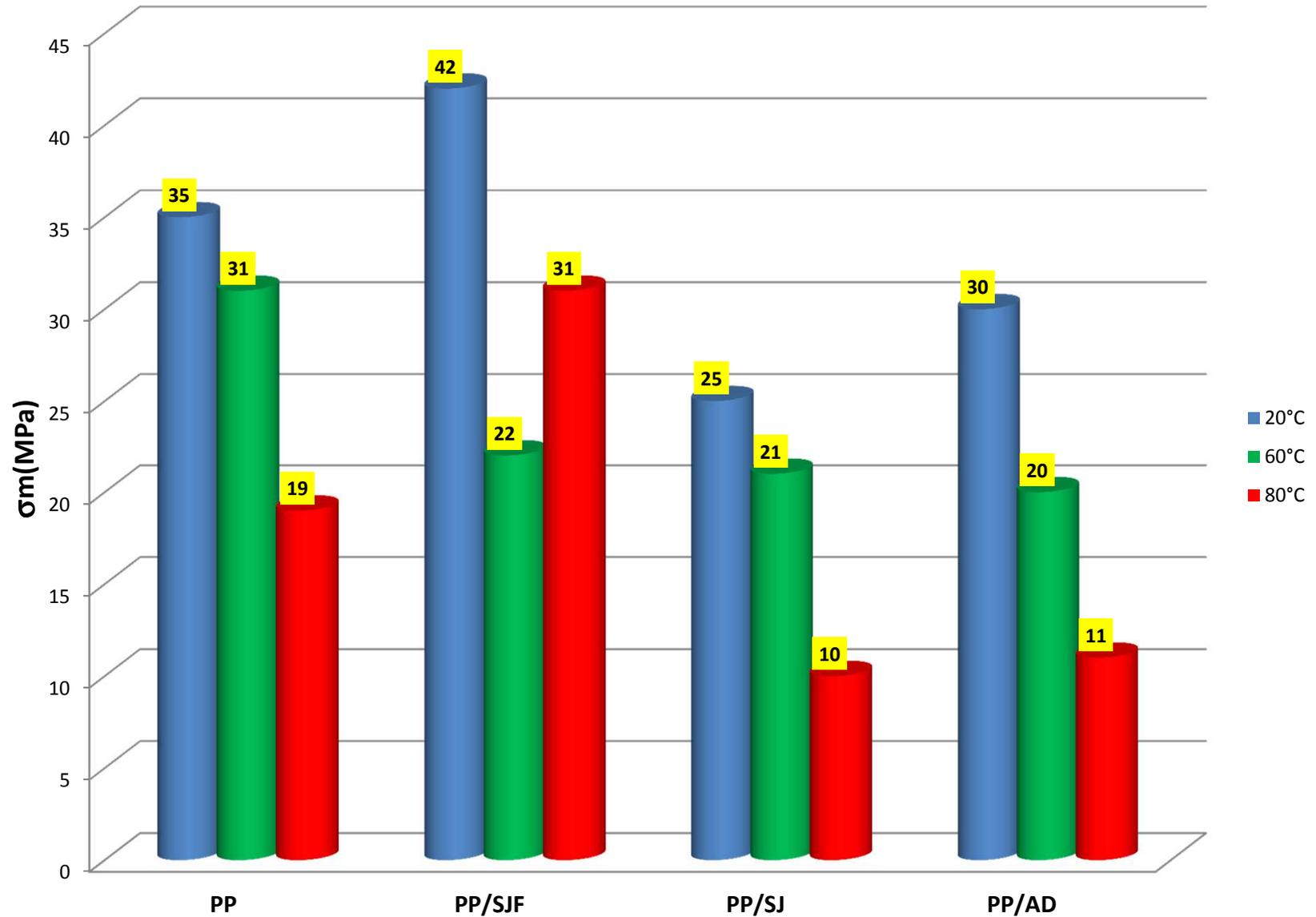


double Extruder for fibers/plastic mixing

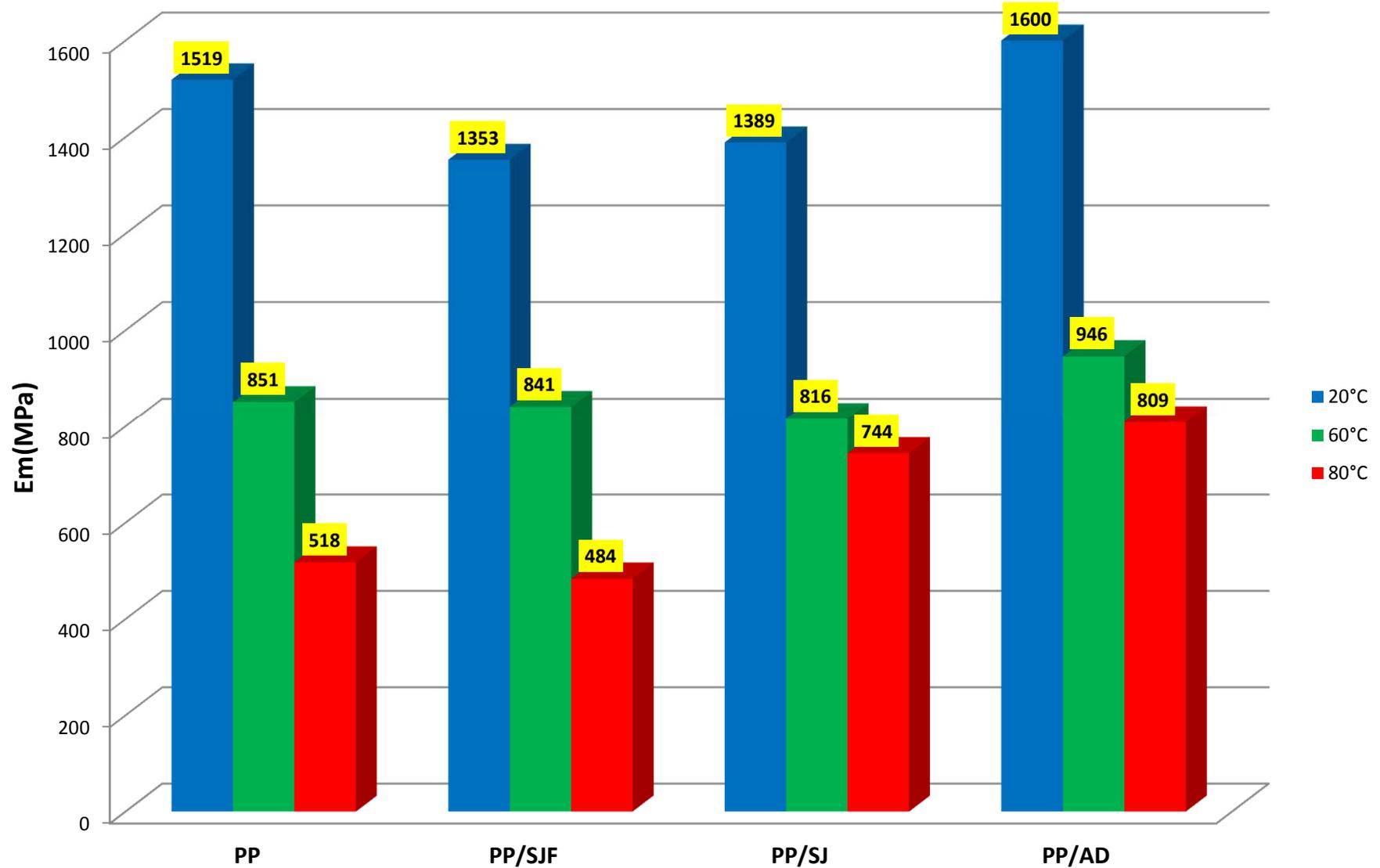
PP composites with With broom and reed fibers



Flexural strength weight ratio (PP/fibres) = 9

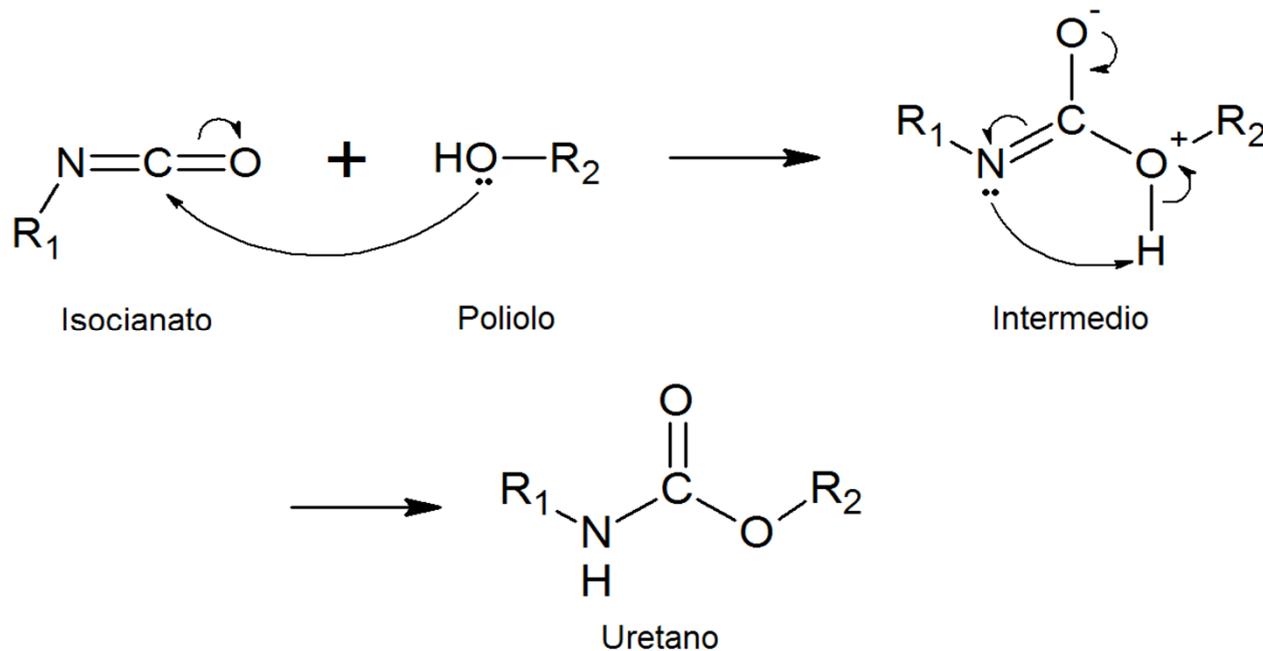


flexural elastic modulus
weight ratio (PP/Fibres) = 9

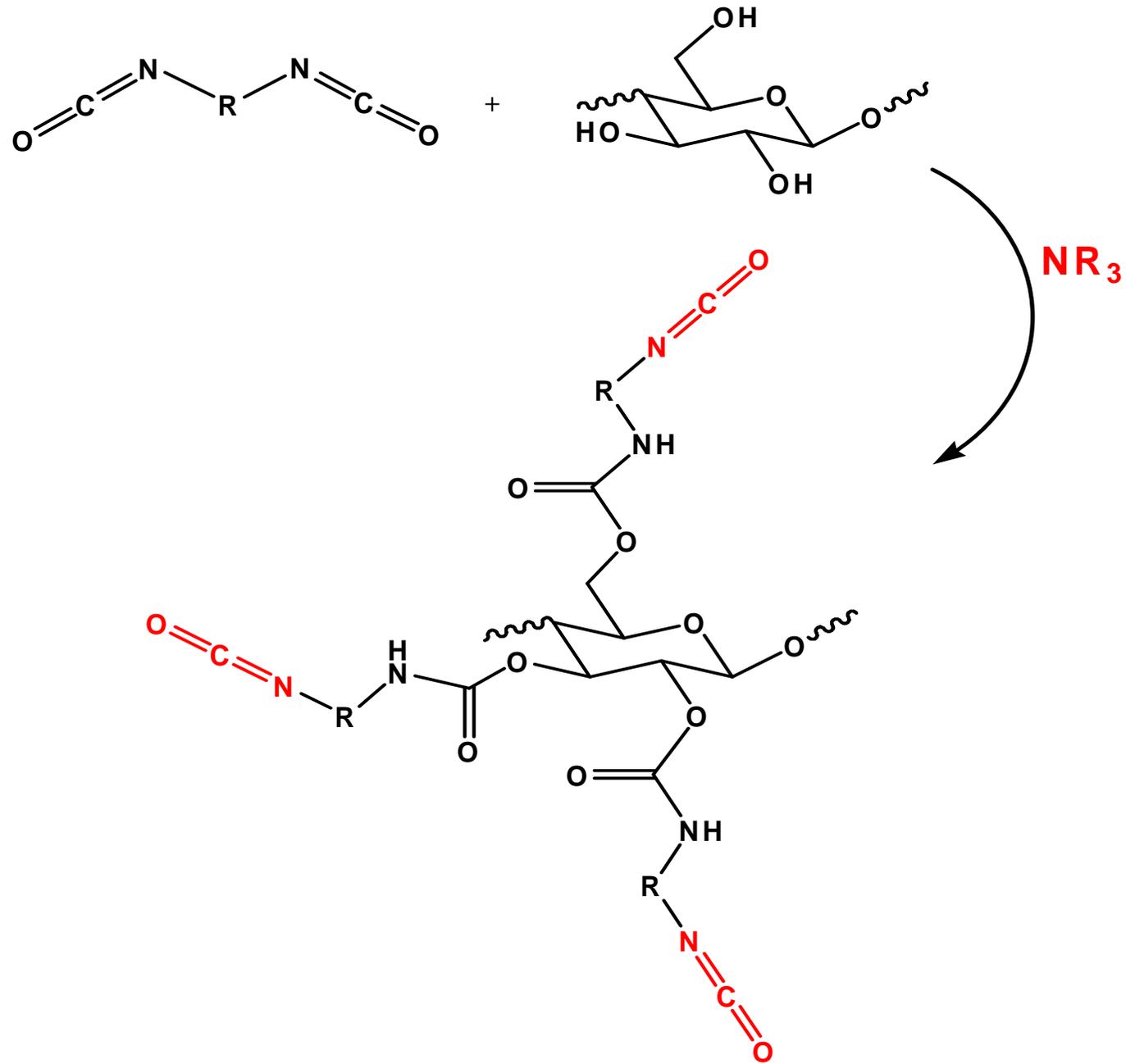


Sintesi di poliuretani da fibra di ginestra micronizzata

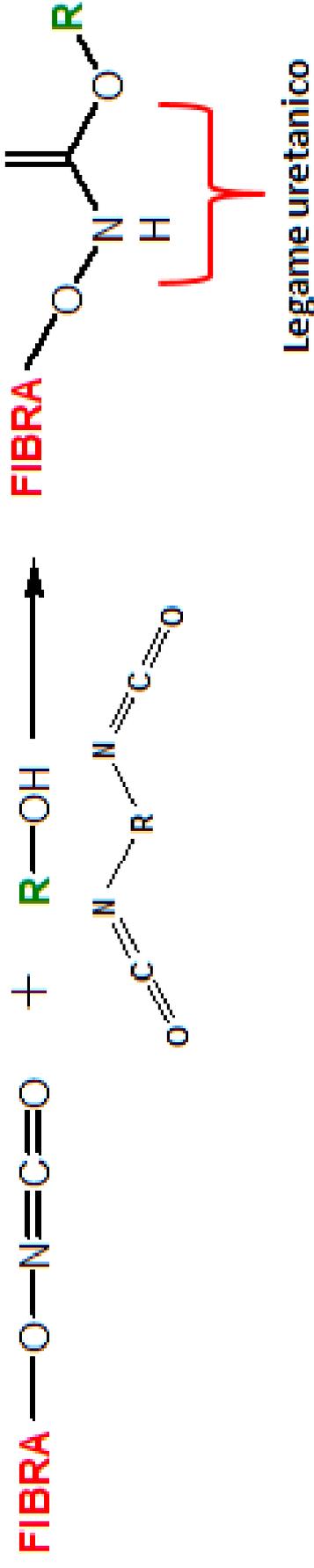
La chimica dei PU è essenzialmente quella degli isocianati, del gruppo NCO e della sua elevata attitudine a reagire con gruppi ossidrilici:



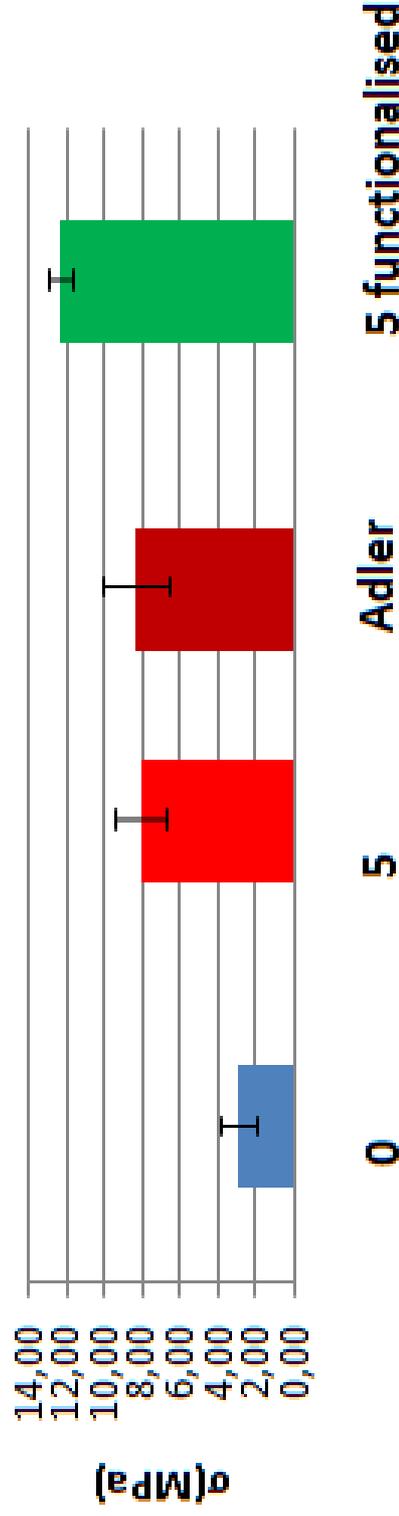
Reazione tra fibra e diisocianato:



Formazione del poliuretano:

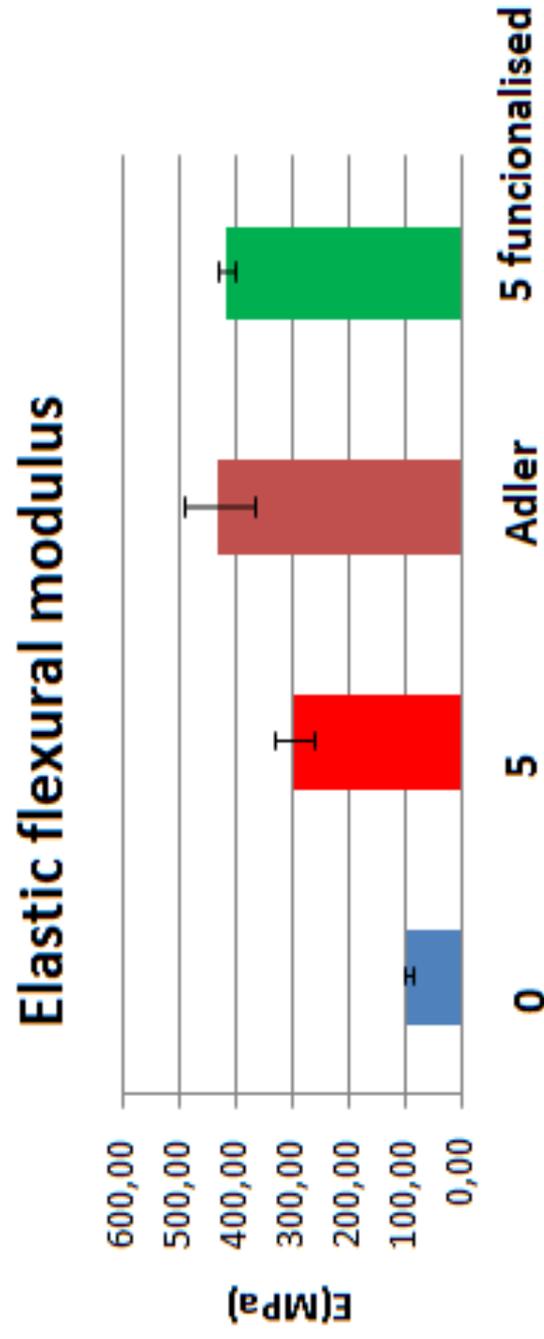


flexural strength



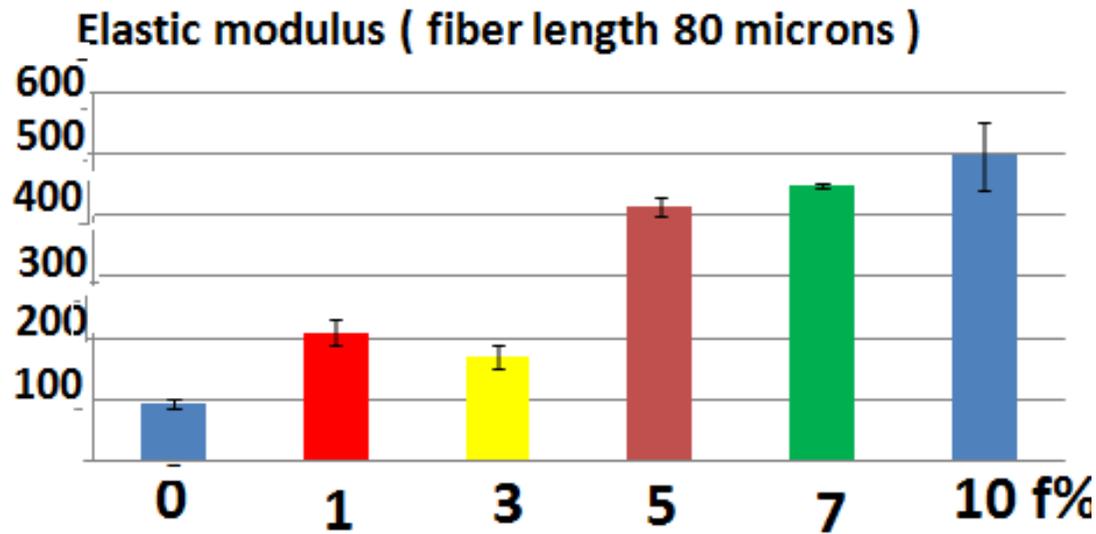
orizzontale axis reports the percentage of fiber

Adler= PU produced by ADLER with mineral fibers

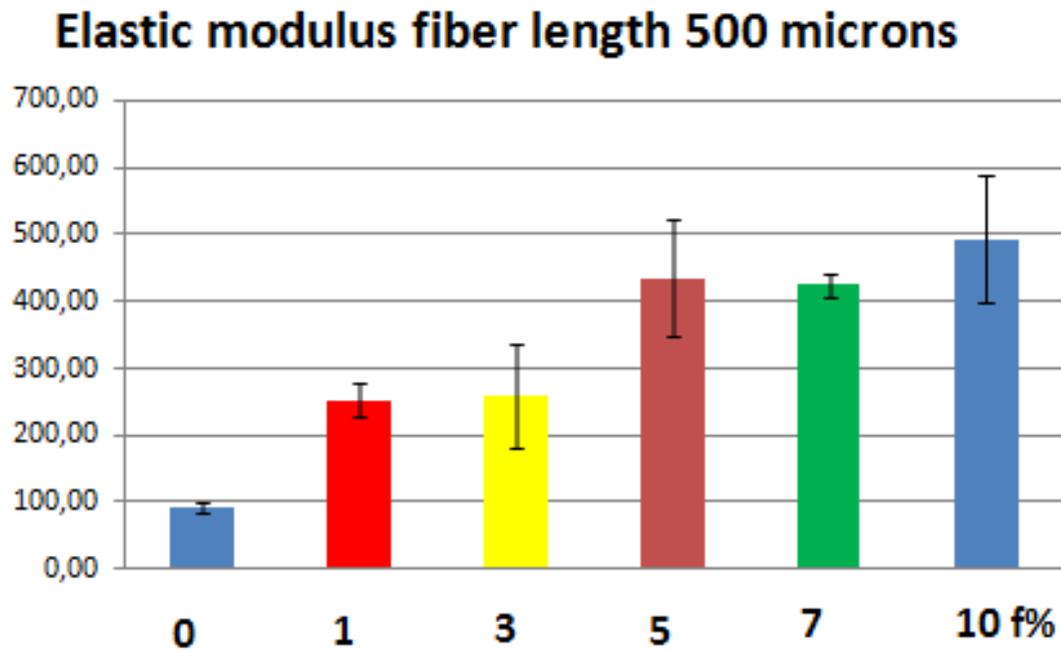


orizzontale axis reports the percentage of fiber

Adler= PU produced by ADLER with mineral fibers

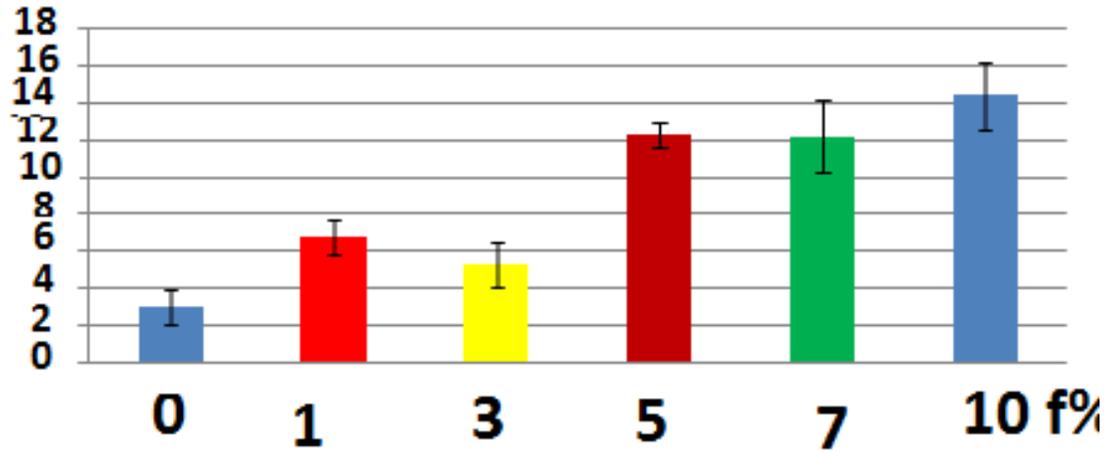


Flexural elastic modulus
For PU/functionlised
cellulose fibers



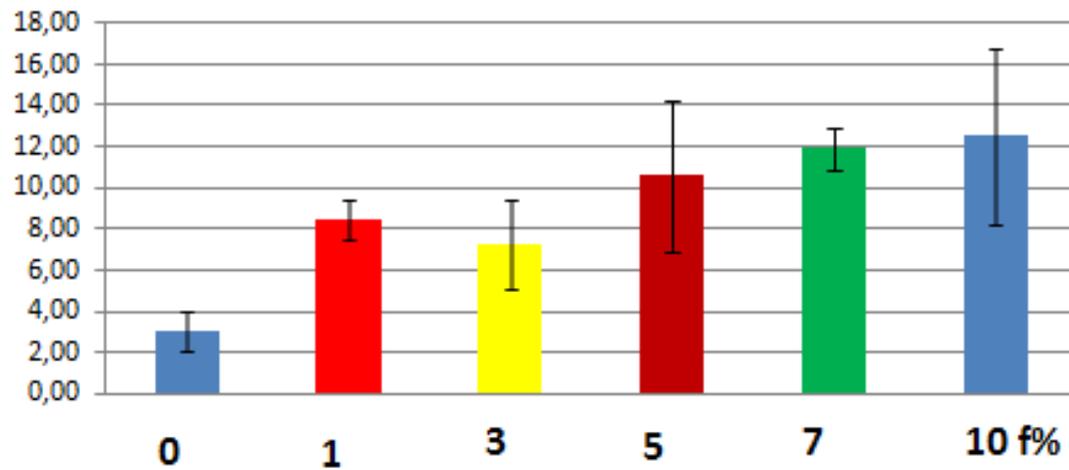
Vertical axis in MPA

flexural strength (fiber length 80 microns)



Flexural strength
For PU/functionlised
cellulose fibers

flexural strength fiber lenght 500 microns

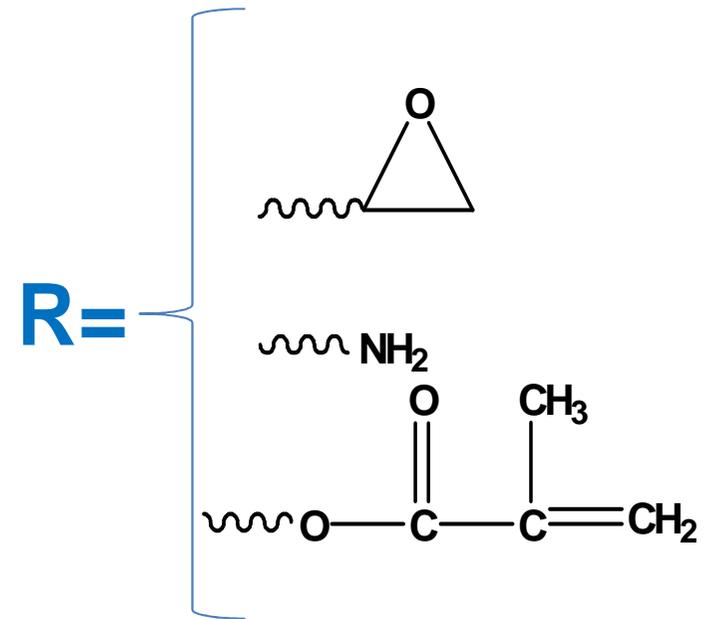
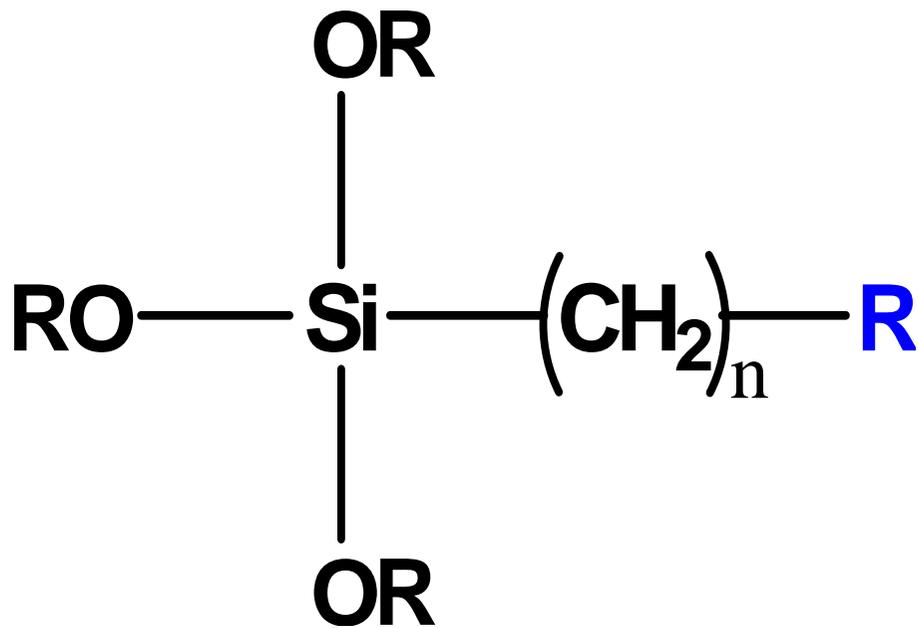


Vertical axis in MPA

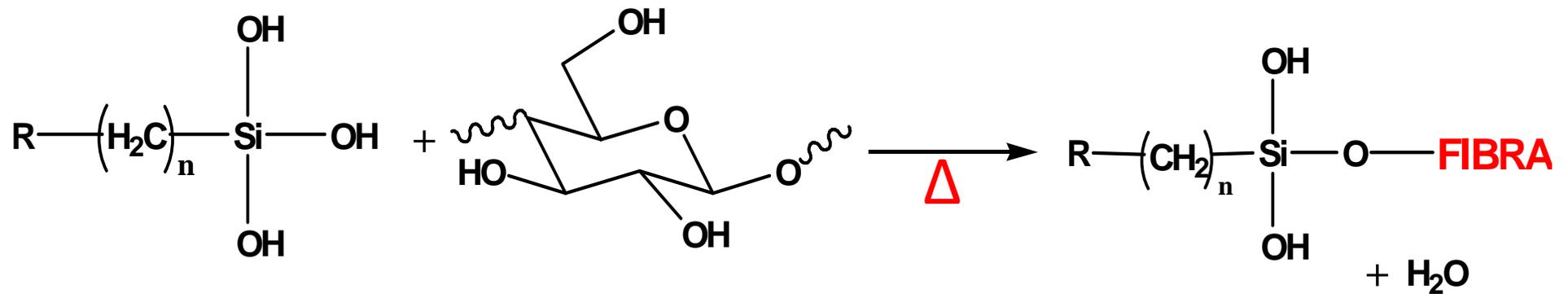
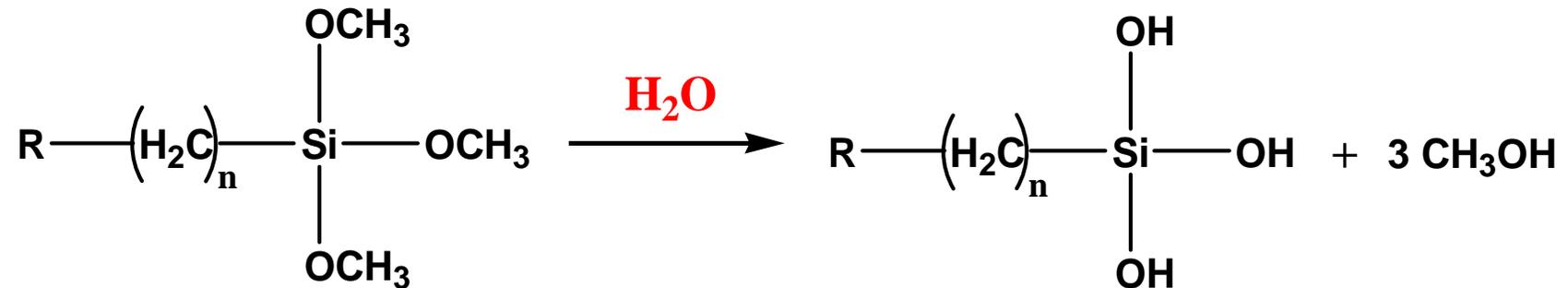
NUOVI COMPATIBILIZZANTI IN VIA DI SPERIMENTAZIONE



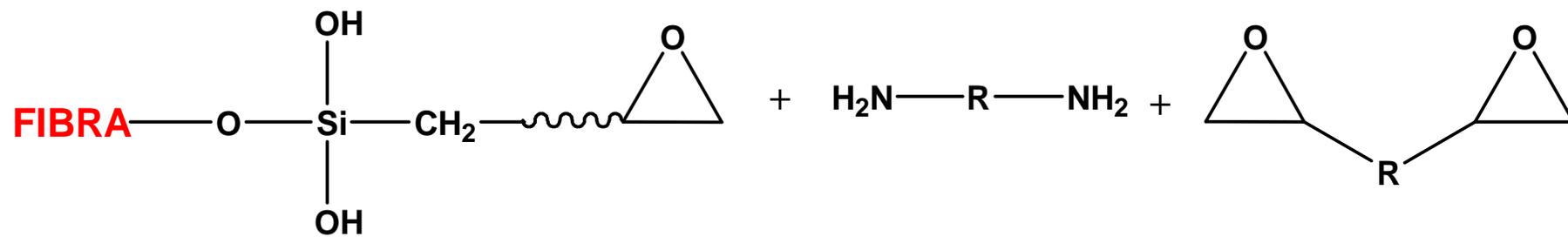
SILANI



I silani reagiscono con i gruppi ossidrilici della fibra previa idrolisi dei gruppi OR legati all'atomo di silicio i quali possono essere gruppi -metossi o -etossi:

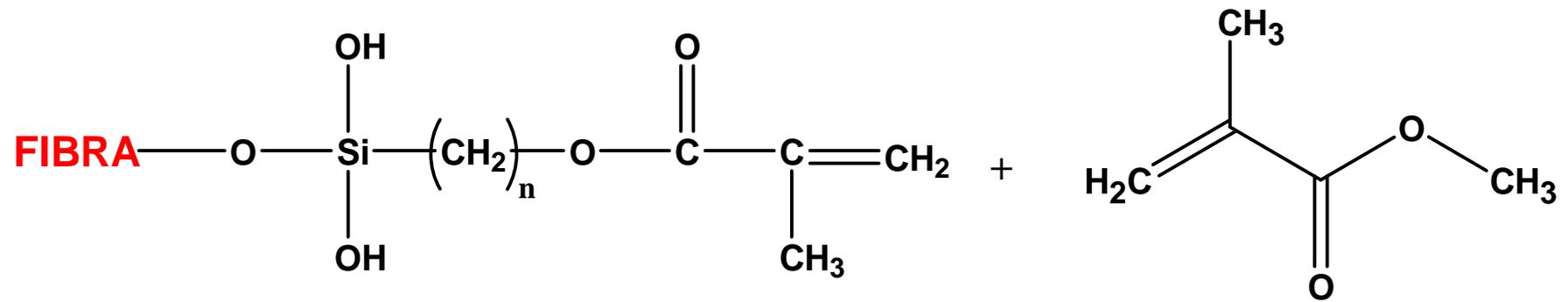


Inizialmente si formano legami idrogeno tra i gruppi ossidrilici che ad alte temperature diventano covalenti.



↓
reticolazione

POLIMERO



PEROSSIDO

Polimerizzazione radicalica

POLIMERO